

SYNTHESIS OF ISOCOFORMYCIN, AN
ADENOSINE DEAMINASE INHIBITOR
OF SYNTHETIC ORIGIN

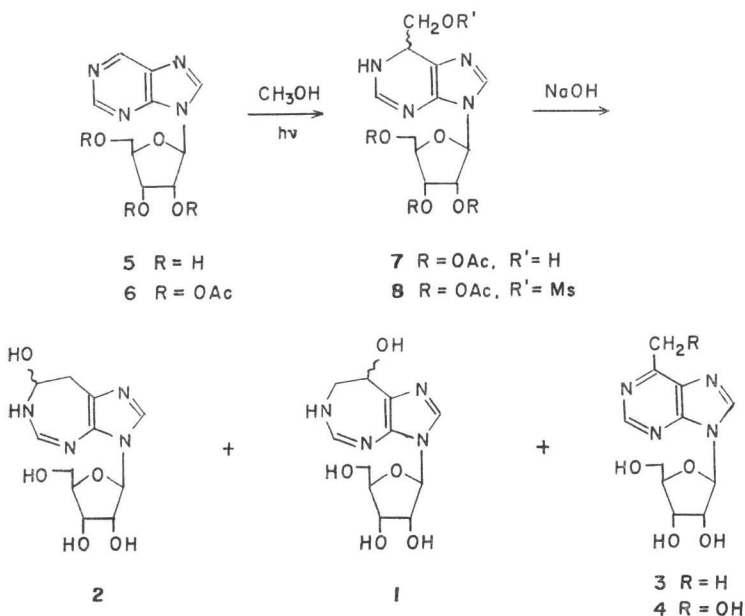
Sir:

Since the structure and the interesting biological activity of coformycin (**1**) were disclosed^{1,2}, a considerable amount of attention has been focused on such unusual nucleosides of microbial origin³⁻⁷. Such modification of the base moiety by nature has led us to approach various seven-membered nucleosides by chemical synthesis and to investigate structure-activity relationships of such unusual nucleosides. In this communication, we describe a synthesis of an isomer of coformycin, 3- β -D-ribofuranosyl-3,6,7,8-tetrahydroimidazo-[4,5-d][1,3]-diazepin-7-ol (**2**) named isocoformycin, which has also been found to possess significant biological activity.

Previously we reported a synthesis of coformycin starting from 9- β -D-ribofuranosylpurine (**5**) or nebularine, but later it became apparent that the products formed in each step were very sensitive to reaction conditions, requiring more careful analysis such as with tlc and high pressure liquid chromatography. The present approach also centered on two key synthetic operations: (a) photoaddition of methanol to **5** and (b) ring expansion of a major product **7** to seven-membered compounds.

Thus, the photoaddition of methanol to **6** was able to be carried out in a stereoselective manner by using Vycor filter (30 W low-pressure mercury lamp, 0~5°C, Argon atmosphere), predominantly affording one of the diastereomers of **7** (EtOAc-MeOH, 5:1; Rf 0.10, 8% and Rf 0.20, 75%)*. The major adduct (mp 58~61°C, $[\alpha]_D^{25} -30.0^\circ$ (c 3.0, MeOH)) was separated by silica gel chromatography (benzene-acetone eluent) and subjected to mesylation with mesyl chloride in the presence of potassium carbonate.

The unstable mesylate (**8**) was immediately treated with dilute sodium hydroxide in aqueous 1,2-dimethoxyethane at room temperature for 2 hours. After neutralizing the solution with acetic acid, the solution was subjected to the following column chromatography using ion exchange resins. Treatment with Amberlite IR-120 resin (NH₄⁺: H⁺=1:1, v/v) and development with 1 N NH₄OH afforded fractions having λ_{max} at 280 nm. The condensed fractions were again treated with Amberlite CG-50 resin (Type I, NH₄⁺: H⁺=1:1, v/v) and eluted simply with water. Four products were obtained from the fractions. 9- β -D-Ribofuranosyl-6-methylpurine (**3**)⁹⁻¹⁰, 9- β -D-ribofuranosyl-6-hydroxymethylpurine (**4**)¹¹ and an epimeric mixture of coformycin were obtained in 20, 10, and 6% yields, respectively. The fourth compound was obtained in 30% yield from the last fractions and proved to be 3- β -D-ribofuranosyl-3,6,7,8-tetrahydroimidazo-[4,5-d][1,3]-diazepin-7-ol named isocoformycin based on the chemical and spectroscopic evidences. ¹H-NMR and ¹³C-NMR are consistent with structure **2**, but ¹³C-NMR shows doublet of each peak demonstrating that **2** is an epimeric mixture of alcohol at C-7. Isoformycin **2** has the formula C₁₁H₁₆N₄O₅ (M⁺ 284); mp 185~190°C; $[\alpha]_D^{25} -58.0^\circ$ (c 1.0, H₂O); UV_{max} (H₂O)



* The photoaddition is very sensitive to reaction conditions and the details of this interesting

photoaddition will be mentioned in a separate paper.

280 nm (ϵ 8200). The hydroxyl group at C-7 is slowly replaced by a methoxyl group in methanol at room temperature and regenerated in hot water, which is characteristic of α -hydroxyamino group¹²⁾. Amazingly, isocoformycin is also a strong inhibitor to adenosine deaminase which will be the subject of a subsequent paper, showing the biological significance of seven-membered nucleosides*.

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* The reaction products containing coformycin and isocoformycin were carefully checked by high pressure liquid chromatography (detector 280 nm UV. Jascorex CV-01 cation exchange resin).